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APPLICATION NO.		FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.	
09/997,436		11/29/2001	John Richard Shutt	2001B110	2001B110 3430	
23455	7590	12/23/2003		EXAMINER		
		CHEMICAL COMPA	HOPKINS, ROBERT A			
POBOX 2 BAYTOW		77522-2149		ART UNIT	PAPER NUMBER	
	.,			1724		
				DATE MAILED: 12/23/2005	3	

Please find below and/or attached an Office communication concerning this application or proceeding.

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	Application	No.	Applicant(s)	7					
,	09/997,436	;	SHUTT ET AL.	φ					
Office Action Summary	Examiner		Art Unit						
	Robert A Ho	<u> </u>	1724						
The MAILING DATE of this communication ap Period for Reply	pears on the d	cover sheet with the c	orrespondence addr	ess					
A SHORTENED STATUTORY PERIOD FOR REPL THE MAILING DATE OF THIS COMMUNICATION. - Extensions of time may be available under the provisions of 37 CFR 1. after SIX (6) MONTHS from the mailing date of this communication. - If the period for reply specified above is less than thirty (30) days, a rep - If NO period for reply is specified above, the maximum statutory period - Failure to reply within the set or extended period for reply will, by statut - Any reply received by the Office later than three months after the mailir earned patent term adjustment. See 37 CFR 1.704(b). Status	136(a). In no event oly within the statuto will apply and will e te, cause the applica og date of this comm	t, however, may a reply be tim ory minimum of thirty (30) days expire SIX (6) MONTHS from a tion to become ABANDONEI	ely filed s will be considered timely. the mailing date of this common (35 U.S.C. § 133).	nunication.					
1) Responsive to communication(s) filed on	-								
2a)☐ This action is FINAL . 2b)⊠ This	action is non	n-final.							
3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under Ex parte Quayle, 1935 C.D. 11, 453 O.G. 213.									
Disposition of Claims									
4) Claim(s) 1-32 is/are pending in the application									
4a) Of the above claim(s) is/are withdrawn from consideration.									
5) Claim(s) is/are allowed.									
6)⊠ Claim(s) <u>1-32</u> is/are rejected. 7)□ Claim(s) is/are objected to.									
8) Claim(s) are subject to restriction and/o	or election rec	quirement.							
Application Papers									
9)☐ The specification is objected to by the Examine	er.								
10)☐ The drawing(s) filed on is/are: a)☐ acc	10) The drawing(s) filed on is/are: a) accepted or b) objected to by the Examiner.								
Applicant may not request that any objection to the	drawing(s) be	held in abeyance. See	37 CFR 1.85(a).						
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).									
11)☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.									
Priority under 35 U.S.C. §§ 119 and 120									
12) Acknowledgment is made of a claim for foreig a) All b) Some * c) None of:	•	•)-(d) or (f).						
1. Certified copies of the priority document2. Certified copies of the priority document			on No						
3. Copies of the certified copies of the prior		• •	· · · · · · · · · · · · · · · · · · ·	age					
application from the International Burea	au (PCT Rule	17.2(a)).							
* See the attached detailed Office action for a list 13) Acknowledgment is made of a claim for domest		•		nnlication)					
since a specific reference was included in the fir		<u> </u>							
37 CFR 1.78. a) ☐ The translation of the foreign language pr	ovisional anni	lication has been rec	eived						
14) Acknowledgment is made of a claim for domest				specific					
reference was included in the first sentence of the									
Attachment(s)									
1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948)	03 62 5	(s) Interview Summary (s) Notice of Informal Pa	(PTO-413) Paper No(s). atent Application (PTO-1						
3) Information Disclosure Statement(s) (PTO-1449) Paper No(s) 1	<u>1-28, 4-15</u> . 6	S) Other: .							

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DETAILED ACTION

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-9 are rejected under 35 U.S.C. 103(a) as being unpatentable over Miller et al(6403854) taken together with Chase et al(4218569).

Miller et al discloses a method of removing oxygenated hydrocarbons from an olefin stream comprising a step of contacting an oxygenate with an olefin forming catalyst to form an olefin stream, wherein the olefin stream comprises olefin and oxygenated hydrocarbon, and contacting the olefin stream with an absorbent in an absorption system to absorb the oxygenated hydrocarbon from the olefin stream(column 14 lines 16-19). Miller et al is silent as to wherein the absorbent is selected from the group consisting of a polyol, amine, amide, nitrile, heterocyclic nitrogen containing compound, and mixtures thereof. Chase et al discloses etherifying an olefin by reacting an olefin with methanol and a catalyst, and using a polyol as an absorbent to remove the methanol. It would have been obvious to someone of ordinary skill in the art at the time of the invention to provide a step of contacting the olefin

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stream of Miller et al with a polyol absorbent because the polyol absorbent is an efficient absorber of the oxygenated hydrocarbon(methanol) while it minimizes absorption of hydrocarbons(column 3 lines 49-52 of Chase et al). Miller et al further teaches removing an olefin stream from the absorption system. Miller et al further teaches wherein the olefin stream removed from the absorption system contains at least 50 wt% less oxygenated hydrocarbon than the olefin stream formed by contacting the oxygenate with the catalyst. Chase et al further teaches wherein the absorbent is ethylene glycol, diethylene glycol, or triethylene glycol. Miller et al further teaches wherein the absorption system is a countercurrent liquid absorption column. Miller et al further teaches compressing the olefin(216) stream prior to contacting the absorbent. Miller et al further teaches contacting the olefin stream recovered from the absorption zone with an absorbent(222) to form an olefin product stream. Miller et al further teaches wherein the olefin product stream. Miller et al further teaches wherein the olefin product stream. Miller et al further teaches

Claims 10-18 are rejected under 35 U.S.C. 103(a) as being unpatentable over Miller et al(6403854) taken together with Chase et al(4218569).

Miller et al discloses a method of removing oxygenated hydrocarbons from an olefin stream comprising providing an olefin stream containing at least 50 wt% ethylene and propylene(column 5 lines 55-57), not greater than 20 wt% water, and not greater than 15 wt% oxygenated hydrocarbon, and contacting the olefin stream with an absorbent to absorb oxygenated hydrocarbon from the olefin stream. Miller et al is silent as to wherein the absorbent is selected from the group consisting of a polyol,

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amine, amide, nitrile, heterocyclic nitrogen containing compound, and mixtures thereof. Chase et al discloses etherifying an olefin by reacting an olefin with methanol and a catalyst, and using a polyol as an absorbent to remove the methanol. It would have been obvious to someone of ordinary skill in the art at the time of the invention to provide a step of contacting the olefin stream of Miller et al with a polyol absorbent because the polyol absorbent is an efficient absorber of the oxygenated hydrocarbon(methanol) while it minimizes absorption of hydrocarbons(column 3 lines 49-52 of Chase et al). Miller et al further discloses wherein the olefin stream contains at least 55 wt% and at least 60 wt% ethylene and propylene. Miller et al further teaches wherein the olefin stream contains not greater than 15 wt% and not greater than 10 wt% water. Miller et al further discloses wherein the olefin stream contains not greater than 12 wt% and not greater than 10 wt% oxygenated hydrocarbon. Chase et al further discloses wherein the absorbent is ethylene glycol, diethylene glycol, and triethylene glycol. Miller et al further teaches polymerizing olefin in the olefin product stream.

Claims 19-28 are rejected under 35 U.S.C. 103(a) as being unpatentable over Miller et al(6403854) taken together with Chase et al(4218569).

Miller et al discloses a method of removing oxygenated hydrocarbons from an olefin stream comprising providing an olefin vapor stream, wherein the olefin vapor stream comprises olefin, water and oxygenated hydrocarbon, and contacting the olefin vapor stream with an absorbent in an absorption system to absorb oxygenated hydrocarbon from the olefin vapor stream. Miller et al is silent as to wherein the absorbent is selected from the group consisting of a polyol, amine, amide, nitrile,

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heterocyclic nitrogen containing compound, and mixtures thereof. Chase et al discloses etherifying an olefin by reacting an olefin with methanol and a catalyst, and using a polyol as an absorbent to remove the methanol. It would have been obvious to someone of ordinary skill in the art at the time of the invention to provide a step of contacting the olefin stream of Miller et al with a polyol absorbent because the polyol absorbent is an efficient absorber of the oxygenated hydrocarbon(methanol) while it minimizes absorption of hydrocarbons(column 3 lines 49-52 of Chase et al). Miller et al further teaches removing an olefin stream from the absorption system, and wherein the olefin stream removed contains not greater than 100wppm water, and wherein the olefin stream removed contains at least 50 wt% less oxygenated hydrocarbon and water than the provided olefin vapor stream. Chase et al further discloses wherein the absorbent is ethylene glycol, diethylene glycol, and triethylene glycol. Miller et al further discloses wherein the absorption system is a countercurrent liquid absorption column. Miller et al further discloses compressing(216) the olefin stream prior to contact with the absorbent. Miller et al further teaches contacting the olefin stream recovered from the absorption zone with an absorbent(222) to form an olefin product stream. Miller et al further teaches polymerizing olefin in the olefin product stream. Miller et al further teaches wherein the olefin product stream contains not greater than 1 wppm water.

Claims 29-32 are rejected under 35 U.S.C. 103(a) as being unpatentable over Miller et al(6403854) taken together with Chase et al(4218569).

Miller et al discloses a method of removing water from an olefin stream comprising contacting an oxygenate with an olefin forming catalyst to form an olefin

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stream, wherein the olefin stream comprises olefin and water, and contacting the olefin stream with an absorbent in an absorption system to absorb the water from the olefin stream. Miller et al is silent as to wherein the absorbent is selected from the group consisting of a polyol, amine, amide, nitrile, heterocyclic nitrogen containing compound, and mixtures thereof. Chase et al discloses etherifying an olefin by reacting an olefin with methanol and a catalyst, and using a polyol as an absorbent to remove the methanol. It would have been obvious to someone of ordinary skill in the art at the time of the invention to provide a step of contacting the olefin stream of Miller et al with a polyol absorbent because the polyol absorbent is an efficient absorber of the water while it minimizes absorption of hydrocarbons (column 3 lines 49-52 of Chase et al). Miller et al further teaches removing an olefin stream from the absorption system. wherein the olefin stream removed contains at least 50 wt% less water than the olefin stream formed by contacting the oxygenate with the catalyst. Chase et al further discloses wherein the absorbent is ethylene glycol, diethylene glycol, and triethylene glycol.

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Any inquiry concerning this communication or earlier communications from the examiner should be directed to Robert A Hopkins whose telephone number is 571-272-1159. The examiner can normally be reached on Monday-Friday 9:00am-4:00pm, alternate Fridays off.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Blaine Copenheaver can be reached on 571-272-1156. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9310.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is 571-272-0987.

Robert A Hopkins Primary Examiner Art Unit 1724

Rah

December 10, 2003